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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/567,443	02/07/2006	Yoshiaki Nagata	80658(47762)	5049
	7590 05/27/201 NGELL PALMER & D	EXAMINER		
P.O. BOX 5587		RIDER, LANCE W		
BOSTON, MA 02205			ART UNIT	PAPER NUMBER
		1618		
			MAIL DATE	DELIVERY MODE
			05/27/2010	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Advisory Action Before the Filing of an Appeal Brief

Application No.	Applicant(s)		
10/567,443	NAGATA ET AL.		
Examiner	Art Unit		

	LANCE RIDER	1618	
The MAILING DATE of this communication appe	ars on the cover sheet with the c	orrespondence add	ress
THE REPLY FILED <u>14 May 2010</u> FAILS TO PLACE THIS APPI	LICATION IN CONDITION FOR AL	LOWANCE.	
1. The reply was filed after a final rejection, but prior to or on application, applicant must timely file one of the following application in condition for allowance; (2) a Notice of Appetor Continued Examination (RCE) in compliance with 37 C periods:	replies: (1) an amendment, affidavit eal (with appeal fee) in compliance v	, or other evidence, w with 37 CFR 41.31; or	hich places the (3) a Request
a) The period for reply expires <u>3</u> months from the mailing date	of the final rejection.		
b) The period for reply expires on: (1) the mailing date of this A no event, however, will the statutory period for reply expire la Examiner Note: If box 1 is checked, check either box (a) or (MONTHS OF THE FINAL REJECTION. See MPEP 706.07(f	dvisory Action, or (2) the date set forth i ater than SIX MONTHS from the mailing b). ONLY CHECK BOX (b) WHEN THE	date of the final rejection	n.
Extensions of time may be obtained under 37 CFR 1.136(a). The date have been filed is the date for purposes of determining the period of extunder 37 CFR 1.17(a) is calculated from: (1) the expiration date of the s set forth in (b) above, if checked. Any reply received by the Office later may reduce any earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL	ension and the corresponding amount on thortened statutory period for reply original than three months after the mailing date	of the fee. The appropria nally set in the final Office	ate extension fee e action; or (2) as
 The Notice of Appeal was filed on A brief in comp filing the Notice of Appeal (37 CFR 41.37(a)), or any exter Notice of Appeal has been filed, any reply must be filed with AMENDMENTS 	nsion thereof (37 CFR 41.37(e)), to	avoid dismissal of the	
3. The proposed amendment(s) filed after a final rejection, be	out prior to the date of filing a brief	will not be entered be	031160
(a) ☐ They raise new issues that would require further cor (b) ☐ They raise the issue of new matter (see NOTE below (c) ☐ They are not deemed to place the application in better	nsideration and/or search (see NOT w);	E below);	
appeal; and/or		gpyg	
(d) ☐ They present additional claims without canceling a converse NOTE: (See 37 CFR 1.116 and 41.33(a)).	corresponding number of finally reje	cted claims.	
4. The amendments are not in compliance with 37 CFR 1.12	21. See attached Notice of Non-Cor	npliant Amendment (I	PTOL-324).
5. Applicant's reply has overcome the following rejection(s):			,
 Newly proposed or amended claim(s) would be all non-allowable claim(s). 		imely filed amendmer	nt canceling the
7. For purposes of appeal, the proposed amendment(s): a) [how the new or amended claims would be rejected is prov The status of the claim(s) is (or will be) as follows: Claim(s) allowed:		be entered and an ea	xplanation of
Claim(s) objected to: Claim(s) rejected:			
Claim(s) withdrawn from consideration: AFFIDAVIT OR OTHER EVIDENCE			
 The affidavit or other evidence filed after a final action, but because applicant failed to provide a showing of good and was not earlier presented. See 37 CFR 1.116(e). 			
9. The affidavit or other evidence filed after the date of filing entered because the affidavit or other evidence failed to o showing a good and sufficient reasons why it is necessary	vercome <u>all</u> rejections under appea	l and/or appellant fail:	s to provide a
 The affidavit or other evidence is entered. An explanation REQUEST FOR RECONSIDERATION/OTHER 	n of the status of the claims after er	itry is below or attach	ed.
The request for reconsideration has been considered but See Continuation Sheet.	t does NOT place the application in	condition for allowan	ce because:
12. ☐ Note the attached Information <i>Disclosure Statement</i>(s). (13. ☐ Other:	PTO/SB/08) Paper No(s)		
/Michael G. Hartley/ Supervisory Patent Examiner, Art Unit 1618			

Continuation of 11. does NOT place the application in condition for allowance because: Applicant argues that it would not be obvious to substitute europium for ytterbium in the dye complexes disclosed by Alburger and Tomita. Applicant alleges that Alburger teaches 783 cordination compounds and teaches that not all metals can be used with every compound, yeilding a Markush of thousands of possible complexes. While Alburger does state that not every metal will form a complex with every compound, Alburger teaches only 7 different fluorescent-chelate-forming beta-diketones, 1-phenyl-1-1, 2-propanedione, 1-pheyl-1,3-butanedione (benzoyl acetone), 1,3-diphenyl-1,3propanedione (dibenzoyl methane), 4,4,4-trifluoro-1-phenyl-1,3-butanedione, 4,4,4-trifluoro-1-(2-napth (2-napth (2-napth yl)-1,3-butanedione, 4,4,4-trifluoro-1-(2-napth yl)-1 trifluoro-1-(2-thienyl)-1,3-butanedione, and 4,4,4-trifluoro-1-(2-furyl-1, 3-butanedione. The substitution of the beta-diketones of Tomita used for forming a flourescent complex for one of the 7 beta diketones disclosed by Alburger used for forming florescent complexes is an obvious substitution as all of the named compounds cordinate the metals in the same way and have the same core structure. Alurger also teaches the following 27 metals aluminum, barium, beryllium, cadmium, calcium, cerium, dysprosium, erbium, europium, gadolinium, gallium, gold, indium, lutecium, magnesium, niobium, ruthenium, samarium, scandium, strontium, terbium, thorium, vanadium, vtterbium, vttrium, zinc, and zirconium. Alburger recites all of the metals as potential equivalents, though he indeed does mention not all of them will work in every complex. Even with no further teachings taken from Alburger the markush of 7 possible beta diketones with every single of the 27 possible metals would yeild a total of only 189 combinations. However, Alburger specifically teaches examples of the lanthanide europium with beta diketones functions to form fluorescent complexes. The other structurally similar metals in the list chemically similar to europium are the tervalent fluorescent lanthanide metals cerium, samarium, gadolinium, terbium, dysporium, erbium, lutecium, and ytterbium. Leaving 7 structurally similar lanthanide metals to replace europium with. This would yeild a markush of 7 beta diketones and 8 metals, or 56 possible compounds. As the method of forming the complexes is merely the mixing the compounds and metals together, there is no significant unpredictability seen in forming any of these 56 complexes. It is still deemed obvious to substitute the lanthanide ytterbium for the structurally similar lanthanide europium in order to obtain a similar fluorescent complex. Further even if applicant contends that the skilled artisan would not substitute structurally similar compounds for one another in such complexes, the markush taught by Alburger contains a total of 183 compounds. Even disregarding the structural similarities of europium and ytterbium it is still considered obvious to select the metal vtterbium from the finite number of identified, predictable solutions, with a reasonable expectation of success.